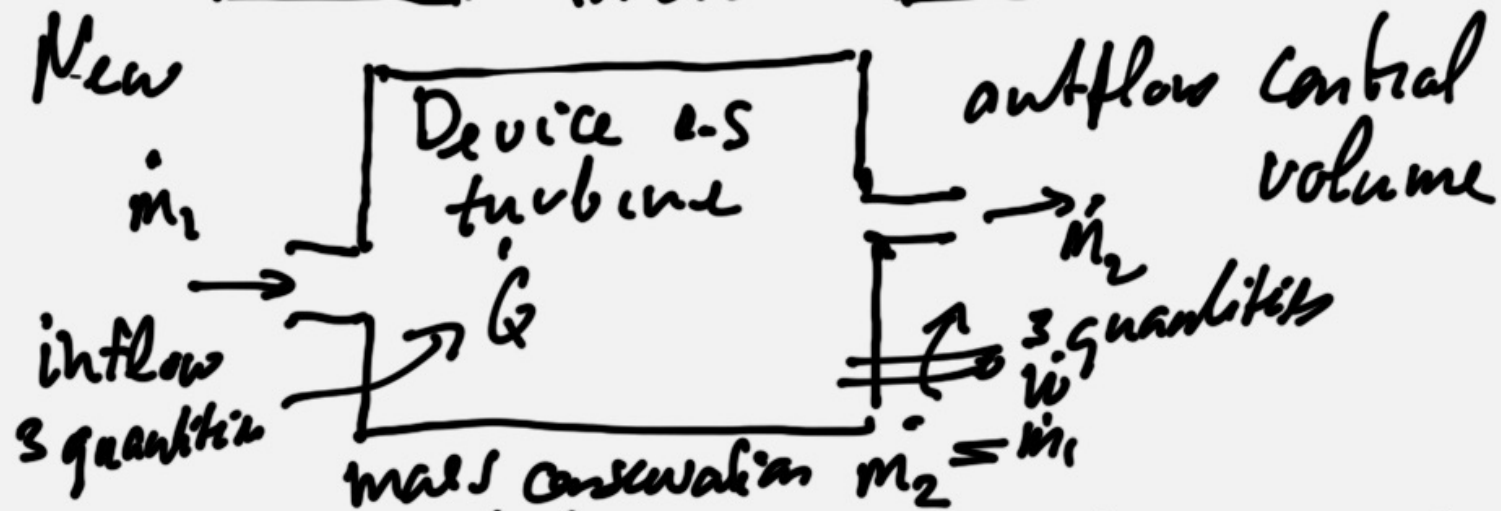
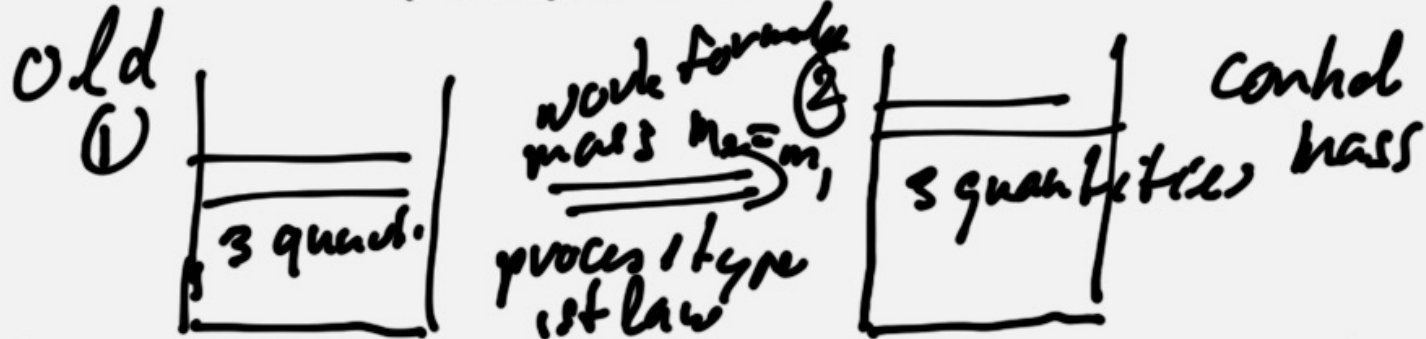


hi3002 Chapter 6 Control volumes



We assume steady state.

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Mass conservation

$$\dot{m}_{in} = \dot{m}_{exit}$$

\dot{m} : mass flow rate

$\dot{m}_1 = \dot{m}_2$ if single entrance and single exit (SEE)

If not SEE: sum the entrances and sum the exits together.

$$\dot{m} = \rho A v$$

$$\dot{m} = \frac{A v \rho}{v}$$

mass flow rate



$$\dot{m} = \frac{\dot{V} \rho}{v}$$

$$\dot{V} = A v$$

specific volume

volumetric flow rate



1st law
old
C.M.

$$\textcircled{1} \left[\begin{array}{l} h_2 - h_1 = Q_2 - W_2 \textcircled{2} \\ \rightarrow Q_2 + h_1 = W_2 + h_2 \end{array} \right]$$

new
C.V.

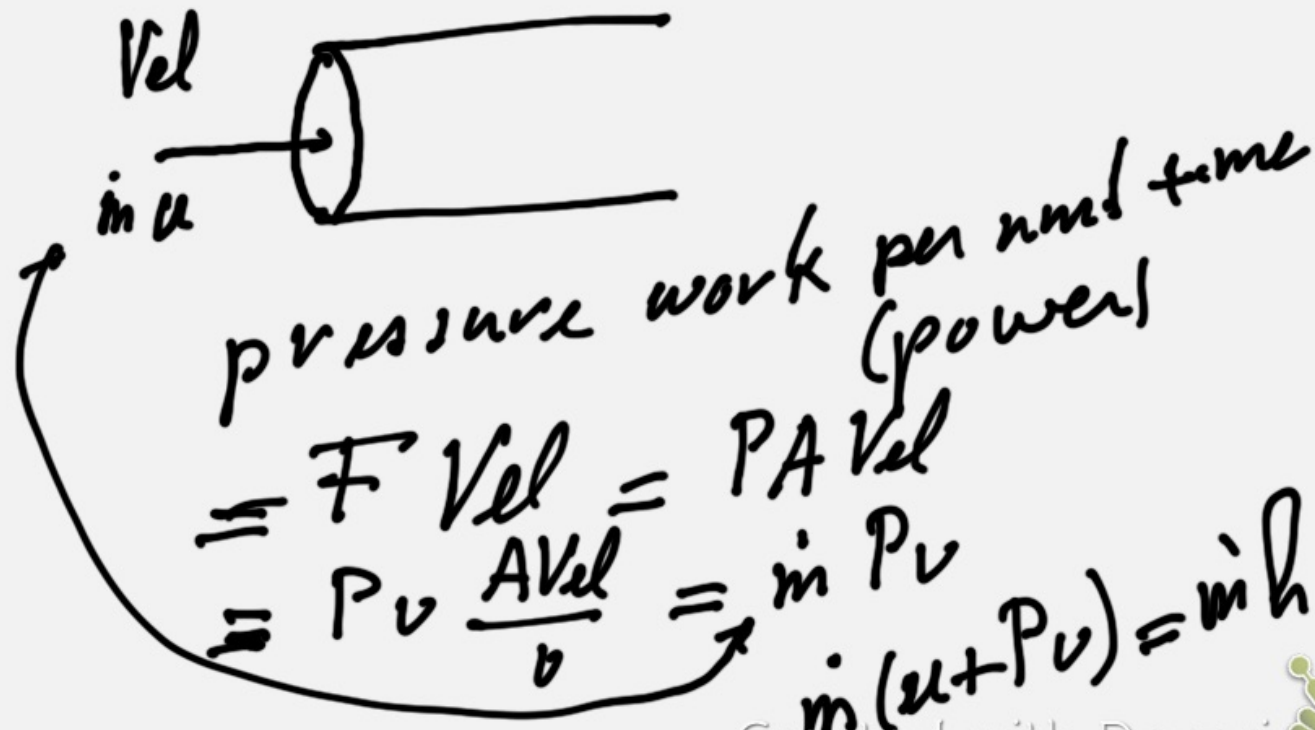
$$\dot{Q} + \dot{m}_1 \left(h_1 + \frac{1}{2} k c_{v1}^2 + g z_1 \right) = \dot{W} + \dot{m}_2 \left(h_2 + \frac{1}{2} k c_{v2}^2 + g z_2 \right)$$

If more than one entrance or exit sum them

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why h instead of u .



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Example: A-6.1

liquid water turbine

specific work w or N power

process type? mass conservation

ignore height differences

2. kJ/s E


100 kPa I
20°C I
Vel ≈ 0 Vel

2000 kPa I
20°C I
15 m/s Vel

Asked: specific work $w = W/m?$
 $\dot{w}?$

1st law is 1 equation for two unknown \dot{Q} and \dot{W} → Table A.1: c.

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Common assumption for turbines is adiabatic
 $\dot{Q} = 0$ ($q = \dot{Q}/\dot{m} = 0$)

On exams, do not make such -
 assumptions, except (besides
 gravity) ~~if it does not have moving~~
 parts, $W = 0$
 e.g. heat exchanger,
 expansion valve

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Solution: at 1, 2000 kPa, 20°C

212.42°C from B.1.2 @ 2000 kPa
 20°C
 \downarrow
 B.1.4 $h_1 = 85.82 \frac{\text{kJ}}{\text{kg}}$

99.62°C
 20°C
 \rightarrow B.1.4 X
 \rightarrow B.1.10 20°C
 \downarrow
 $h_2 = 83.94 \frac{\text{kJ}}{\text{kg}}$

h_1 : C.L. table B.1.4
 h_2 : C.L. table B.1.10

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$$\dot{Q} + \dot{m}_1 (h_1 + \frac{1}{2} Vel_1^2 + \dots) \quad \frac{kg \frac{m^2}{s^2}}{s}$$

$$= \dot{W} + \dot{m}_2 (h_2 + \frac{1}{2} Vel_2^2 + \dots)$$

$$2 (\rho_5 \cdot \rho_2 + \frac{1}{2} 15^2) = \dot{W} + 2 (\rho_3 \cdot 94)$$

$$W = 2 (\rho_5 \cdot \rho_2 - \rho_3 \cdot 94 + \frac{1}{2} 15^2)$$

$$= \underline{\underline{crap}}$$

$$2 \cdot \frac{kg}{s} (\rho_5 \cdot \rho_2 \frac{kg}{s} - \rho_3 \cdot 94 \frac{kg}{s}) + \frac{1}{2} 15^2 \frac{m^2}{s^2} \left(\frac{kg}{1000 \frac{m^3}{s^3}} \right)$$

$$= 3 \cdot \rho \frac{kg}{s} = 3 \rho \text{ kW} \quad W: \text{Watt}$$

$$w = W/\dot{m} = 3 \cdot \rho \frac{kg}{s} / 2 \frac{kg}{s} = 1.5 \frac{kg}{s}$$

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