Usage of Microbial Fuel Cell Technology in Landfills. Year II. Enhanced Organic Compound Decomposition and Nitrogen Removal

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#### Ammonium Source in Landfill Leachate

#### Protein

> 0.5% of Dry Weight of Municipal Solid Waste

### Landfill Leachate with High Ammonium Content







- > Organic Decomposition
- Nutrient Removal
- Energy Generation

#### Landfills at Isolated Locations

















NO2

### **Anammox for Ammonium Removal**

**O**<sub>2</sub>





O2

## **Electricity Generation**



Electron Consumption Separated from Organic Carbon Oxidation

> Landfill Leachate organic Compound Decomposition





 $C_{12}H_{22}O_{11} + 13H_2O \rightarrow 12CO_2 + 48H^+ + 48e^-$ 

Electricity Generation



### **Prior Research – Microbial Fuel Cell**







### **Power Generation — Batch MFC**



### **Power Generation — Continuous MFC**



### **Organic Decomposition**



## **Organic Decomposition**

	<b>K</b> <sub>S</sub> ( <b>mg/L</b> )	Y (g/g)	$\mu_{max}$ (day <sup>-1</sup> )
Glucose	154.3	0.678	0.0124
Gadsden County	271.6	0.323	0.0072
Leon County	172.1	0.412	0.0089
Okaloosa County	163.7	0.486	0.0105
Santa Rosa County	174.5	0.421	0.0093

#### Iron Release in NW Florida



#### Central Landfill Walton County



Fairgrounds Branch below Auto Shred Landfill

#### Visible Iron Release nearby Landfills





### **Iron Release Prevention Experiments**

### Parallel Control Experiments







### **Iron Release Prevention Results**







### **Prior Research – Anammox**







#### **Prior Research – Anammox**



### **Anammox Impacted by Alkalinity**



#### **Anammox Impacted by Partial Nitrification**







## **Objectives**

- 1. Landfill Leachate Treatment in Ammonium Oxidation/MFC Reactor
- 2. Landfill Leachate Treatment in MFC/Anammox Reactor
- 3. System Comparison in terms of Power Generation as well as Organic Compound Decomposition and Nitrogen Removal





### Landfill Soil and Leachate Collection



Leachate was collected from a tank, using a bailer at Franklin County Central Landfill. The leachate tank is located near Monitoring Well MW-19.





## Landfill Leachate Characterization

- > BOD<sub>5</sub> up to 20,000 mg/l
- >  $NH_4^+$ -N up to 500 mg/l
- Phosphorus up to 200 mg/l







Santa Rosa County Central Landfill

#### S. putrefaciens and G. metallireducens Culturing



polymerase chain reaction





- Culturing Media
- ➤ KH<sub>2</sub>PO<sub>4</sub> 160 mg/l
- ≻ K<sub>2</sub>HPO<sub>4</sub> 420 mg/l
- ➢ Na₂HPO₄ 50 mg/l
- ➢ NH₄CI 40 mg/I
- ➢ MgSO₄·7H₂O 50 mg/l
- ➤ CaCl<sub>2</sub> 50 mg/l
- FeCl<sub>3</sub>·6H<sub>2</sub>O 0.5 mg/l
- ➢ MnSO₄·4H₂O 0.05 mg/l
- ≻ H<sub>3</sub>BO<sub>3</sub> 0.1 mg/l
- ➤ ZnSO<sub>4</sub>·7H<sub>2</sub>O 0.05 mg/l
- ➤ (NH4)<sub>6</sub>Mo<sub>7</sub>O<sub>24</sub> 0.03 mg/l
- Glucose 200 mg/l





## **Ammonium Oxidation/MFC Reactor**







Graphite Rod without Catalyst Coating as Anode, Inoculated with S. putrefaciens

Carbon Cloth as Cathode, Inoculated with G. metallireducens

Synthetic Polymeric Nanoporous Membranes as Cation-Exchange Membrane

Treated Leachate Looped into Cathodic Chamber





# **Operation Conditions**

- ➢ Organic Load (up to 20,000 BOD₅)
- Retention Time (up to 80 hrs)
- ≻ pH (4 to 12)
- > Alkalinity (up to 500 mg/l as  $CaCO_3$ )
- Nitrification Reaction Time (up to 2 hrs)





### **MFC/Anammox Reactor**







Graphite Rod without Catalyst Coating as Anode, Inoculated with S. putrefaciens

Carbon Cloth as Cathode, Inoculated with G. metallireducens and Anammox Consortia

Synthetic Polymeric Nanoporous Membranes as Cation-Exchange Membrane

Treated Leachate Looped into Cathodic Chamber

Partial Nitrification





# **Operation Conditions**

- ➢ Organic Load (up to 20,000 mg/I BOD₅)
- Retention Time (up to 80 hrs)
- ≻ pH (4 to 12)
- > Alkalinity (up to 500 mg/l as  $CaCO_3$ )
- Dissolved Oxygen Concentration (up to 0.60 mg/l)





#### **Expected Results**



#### **Expected Results**



#### **Power Generation**

#### **Current and Power**



I: Current

E: Voltage

R: Electrical Resistance

P: Power Output





# **Organic Decomposition Modeling**

### **Organic Decomposition in MFCs**



# $W(x) + \ln[W(x)] = \ln(x)$

S: Organic Compound Concentration  $K_m$ : Half Saturation Constant  $\mu_{max}$ : Maximum Bacterial Growth Rate t: Time

S<sub>0</sub>: Initial Organic Compound Concentration W: Lambert Function





## Nitrogen Removal Modeling

#### **Ammonium Removal Rate**

$$\frac{d[NH_4^+]}{dt} = -\frac{k_r[NH_4^+]}{K_s + [NH_4^+] + [NH_4^+]^2 / K_I} [X]$$

[NH<sub>4</sub><sup>+</sup>]: Ammonium Concentration t: Time (h)

[X]: Bacterial Cell Concentration

K<sub>r</sub>: Ammonium Removal Constant

K<sub>s</sub>: Half Saturation Constant

K<sub>I</sub>: Inhibition Constant





## **System Comparison**

Ammonium Oxidation/MFC Reactor

- Complete Ammonium Oxidation
- No Cation Passing through Cation Exchange Membrance
- MFC/Anammox Reactor
  - Ammonium Oxidation Coupled with Denitrification
  - More Power Generation
  - Complicated





# **Timeline of Milestones**

Activity	Month 1-3	Month 3 - 6	Month 7 – 9	Month 9 - 12
Landfill Leachate and Soil Sample Collection				
S. putrefaciens and G. metallireducens Culturing				
Landfill Leachate Treatment in Ammonium Oxidation/MFC Reactor				
Landfill Leachate Treatment in MFC/Anammox Reactor				
System Comparison				
Reporting	Quarterly Report	Quarterly Report	Quarterly Report	Final Report





### **Ammonium Oxidation/MFC Reactor Setup**



### **Ammonia Depletion**



### **Nitrate Production**



Time (hr)

### **Questions?**









