Introduction to Process Design

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- Planning and Organizational Design
 - Development of design team
 - Budget and time line



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- Utilize engineering judgment to analyze the solution.

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It is desired to produce 150,000 cubic meters of 190 proof ethanol per year from a feed of 75 million kg/year of ethylene. The ethylene feed is 96 mole % ethylene, 3 mole % propylene and 1 mole % methane. The feed costs \$0.18 per pound and it is estimated that ethanol can be sold for \$2.55 per gallon.

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Fundamental Rule of Economics: SELLING PRICE SHOULD BE GREATER THAN COST PRICE

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(because mol. wt. of ethanol is 46.07 and mol. wt. of water is 18.02)

Weight fraction of ethanol is: $\frac{(0.8544)(46.07)}{41.99} = 0.937$ Density of 190 proof ethanol is 810 kg/m^3 . Thus, moles of ethanol in 150,000 m^3/yr of 190 proof ethanol is: $\frac{(0.937)(150,000)(810)}{46.07} = 2,471,000 \ kmol/yr$ Weight fraction of ethanol is: $\frac{(0.8544)(46.07)}{41.99} = 0.937$ Density of 190 proof ethanol is 810 kg/m^3 . Thus, moles of ethanol in 150,000 m^3/yr of 190 proof ethanol is: $\frac{(0.937)(150,000)(810)}{46.07} = 2,471,000 \ kmol/yr$

Assuming 100% conversion of ethylene to ethanol, to produce 1 mole of ethanol requires 1 mole of ethylene. Thus, to produce 2,471,000 kmol/yr of ethanol, we need: (2,471,000)(28.05) = 69,310,000 kg/yr of ethylene The feed is impure. With 96 moles of ethylene, we are also getting 3 moles of propylene and 1 mole of methane.

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This amounts to the following:

$$\frac{3}{96}(2,471,000)(42.08) = 3,249,000 \ kg/yr \text{ propylene}$$
$$\frac{1}{96}(2,471,000)(16.04) = 412,900 \ kg/yr \text{ methane}$$

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Total feed = $72,980,000 \ kg/yr$ Cost of feed = \$0.18/lb (given)

Thus total feed cost =
$$(72, 980, 000)(2.2046)(0.18)$$

 \approx \$29 million

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Thus, we need a process where equipment cost plus operating cost are less than \$72 million.

Ethylene	Ethanol	Max. Profit
0.18/lb	2.55/gal	\$72 million
0.16/lb	2.04/gal	million
0.18/lb	2.68/gal	million
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0.21/lb	3.08/gal	\$89 million