Gas Absorption with Plate Absorbers

One man's magic is another man's engineering. Lazarus Long

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- This unit operation is modeled as a cascade of equilibrium stages
- The assumption of equilibrium is weak and is corrected by the use of tray efficiencies



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- 2. Number of equilibrium stages (N) for a desired recovery of key component (or vice versa)
- 3. Temperature (T_0) of the absorbing liquid stream
- 4. Flowrate (L_0) of the absorbing liquid stream

Given these 4 specifications, we can derive the mass balance relations The liquid and vapor streams leaving a stage are at

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For each stage, the absorption factor A_i will be different. If we assume a constant absorption factor A_E for all the stages, we get simplified mass balance equations.

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Writing the above balance from tray 1 to tray N relates the vapor flow at the bottom, v_{N+1} to the vapor flow at the top, v_1 and the liquid flow at the top, l_0 .

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$$v_{N+1} = \sum_{i=0}^{N} (A_E)^i v_1 - \sum_{i=0}^{N-1} (A_E)^i l_0$$

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Multiplying both sides of the above equation by $(1 - A_E)$ results in:

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which simplifies the relationship for v_{N+1} to:

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The above equations are valid for any component k in the column.

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Substituting for β_N in terms of A_E , and solving for N results in the Kremser equation:

$$N = \frac{\ln\left\{\frac{l_0^n + (r - A_E)v_{N+1}^n}{l_0^n - A_E(1 - r)v_{N+1}^n}\right\}}{\ln\{A_E\}}$$

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- Overhead column pressure
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- The absorption factor $A_E = 1.4$ (choose this as initial guess) which fixes the liquid flow rate

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Calculate the number of stages from the Kremser equation

$$N = \frac{\ln\left\{\frac{l_0^n + (r - A_E)v_{N+1}^n}{l_0^n - A_E(1 - r)v_{N+1}^n}\right\}}{\ln\{A_E\}}$$

Calculate absorption factors for non-key components

$$A^{k} = \frac{L_{0}}{V_{N+1}} \frac{P}{P_{k}^{0}(T)} \quad k \neq n$$

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Calculate aggregate terms (
 values) for non-key components

$$\beta_{N}^{k} = \frac{\left[1 - (A^{k})^{N+1}\right]}{1 - A^{k}}$$
$$\beta_{N-1}^{k} = \frac{\left[1 - (A^{k})^{N}\right]}{1 - A^{k}}$$

$$v_{1}^{k} = \frac{v_{N+1}^{k}}{\beta_{N}^{k}} + \frac{\beta_{N-1}^{k}}{\beta_{N}^{k}} l_{0}^{k}$$

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 - 3. If too many undesirable components are absorbed, increase T, decrease P, or select a more suitable solvent for absorption.